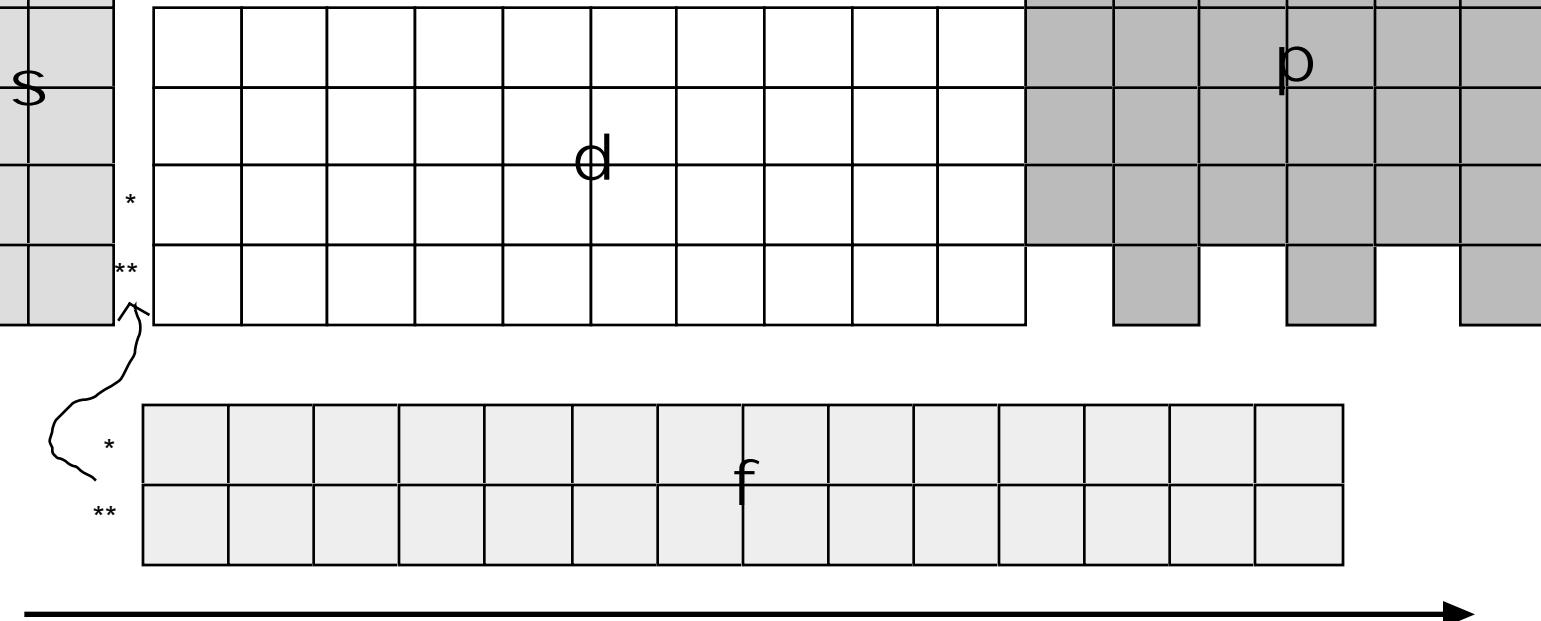
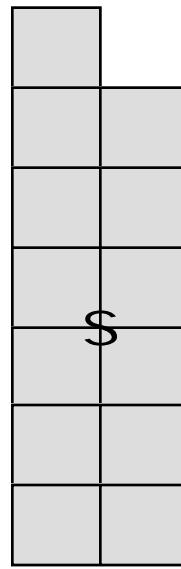


Periodic Trends

Atomic Radius increases
Electronegativity decreases
Shielding effect increases
Ionization Energy decreases



Atomic Radius decreases
Electronegativity increases
Shielding effect is constant
Ionization Energy increases

