## Conversions

1) What is your weight in kilograms?
2) If a canoe can hold 500 kg , how many pounds can it hold?

Name $\qquad$

## Equalities

$1 \mathrm{~kg}=2.2 \mathrm{lbs}$
$1 \mathrm{~atm}=14.69 \mathrm{psi}$
$1 \mathrm{~mol}=6.02 \times 10^{23}$
$1 \mathrm{~g}=1000 \mathrm{mg}$
1 drop of water $=0.10 \mathrm{ml}$
1 mol of $\mathrm{H}_{2} \mathrm{O}$ molecules $=18.0 \mathrm{~g}$
3) Is a tire that is marked "inflate to 35 psi" over or under inflated if the actual pressure is 0.756 atm ? Show work.
4) How many atoms are there in 0.00045 moles of atoms?
5) How many grams would 3.56 moles of carbon atoms weigh?
6) Many cans of soda have 35 mg of sodium. How many grams of sodium is this? How many atoms of sodim is this?
7) (check plus if you can get this one) How many water molecules are in one drop of water? (Hint: you need the density formula in one of the several steps to solving this problem. Density of water $=1.00 \mathrm{~g} / \mathrm{ml}$ )

